					t)					
		ACAD	EMIC SESSION : 2020	-21						
	SAMPI	LE TEST PAPE	R (For XII to XIII Mov	ing, Main Patteri	n)					
Duratio	n : 18 Min.			Max. M	arks : 96					
Name :		Арр	lication Form Number							
		Reg. Number : [	20							
-		GE	NERAL INSTRUCTIONS	)						
1. 2. 3.	<ul> <li>There are 06 pages in the booklet containing 24 questions of Physics (1 to 8), Chemistry (9 to 16), Mathematics(17 to 24) each question carries 4 mark.</li> <li>Blank papers, clip boards, log tables, slide rule, calculators, mobile or any other electronic gadgets in any form is not allowed.</li> <li>Write your Name and Roll No. in the space provided at the top of this booklet.</li> <li>Before answering the paper, fill up the required details in the blank space provided in the answer sheet.</li> <li>Do not forget to mention your roll number neatly and clearly in the blank space provided in the answer sheet.</li> <li>No rough sheets will be provided by the invigilators. All the rough work is to be done in</li> </ul>									
4. 5. 6.	answer shee Do not forg the answer No rough s	et. get to mention your sheet. heets will be provid	roll number neatly and clearly ded by the invigilators. All t	y in the blank space p	provided in					
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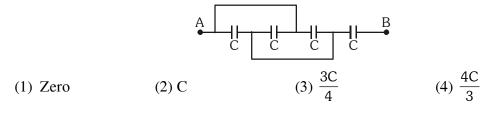
## PHYSICS

This section contains 8 multiple choice questions. Each question has four choices (1), (2), (3) and (4) out of which only one is correct

1. A 10m long potentiometer wire has a potential gradient of 0.0025 V/cm. Calculate the distance of null point when the wire is connected to a 1.025V standard cell :-

(1) 4.1 m (2) 0.25 m (3) 1.0 m (4) 2.0 m

2. The equivalent capacitance of circuit as shown in figure is-



3. A square frame is made from four identical rods each having mass m and length  $\ell$ . Moment of inertia of the frame about one side will be equal to–

(1)  $\frac{3}{2}m\ell^2$  (2)  $\frac{2}{3}m\ell^2$  (3)  $\frac{5}{3}m\ell^2$  (4)  $\frac{11}{3}m\ell^2$ 

4. The length of a given cylindrical wire is increased by 100%. Due to the consequent decrease in diameter the change in the resistance of the wire will be(1) 300% (2) 200% (3) 100% (4) 50%

5. Orbital velocity of a satellite moving in orbit of radius 4R from centre will be-

(1) 
$$\sqrt{\frac{GM}{3R}}$$
 (2)  $\sqrt{\frac{GM}{2R}}$  (3)  $\sqrt{\frac{GM}{4R}}$  (4)  $\sqrt{\frac{3GM}{R}}$ 

6. Magnetic field at point 'M' of given current distribution :-

(1) 
$$\frac{3\mu_0}{\pi} \otimes$$
 (2)  $\frac{5\mu_0}{2\pi} \otimes$  (3)  $\frac{2\mu_0}{\pi} \otimes$  (4)  $\frac{\mu_0}{2\pi} \otimes$ 

7. A train of 150 m length is going towards east with speed 10 m/s. A bird flies at a speed of 5 m/s towards west. The time taken by the bird to cross the train is-

- 8. Stationary wave is represented by  $y = A \sin(100 t) \cos(0.01x)$  where y and A are in mm, t in sec and x in m. The velocity of the wave–
  - (1) 1 m/s (2)  $10^2$  m/s (3)  $10^4$  m/s (4) Not derivable

**R-NET (2020)** 



### **CHEMISTRY**

This section contains 8 multiple choice questions. Each question has four choices (1), (2), (3) and (4) out of which only one is correct

9. The reaction  $2X \rightarrow B$  is a zeroth order reaction. If the initial concentration of X is 0.2 M, the half-life is 6 h. When the initial concentration of X is 0.5 M, the time required to reach its final concentration of 0.2 M will be :-

(1) 18.0 h (2) 7.2 h (3) 9.0 h (4) 12.0 h

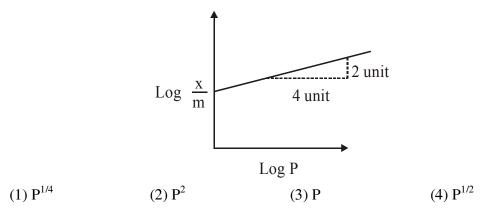
**10.** The standard reaction Gibbs energy for a chemical reaction at an absolute temperature T is given by

$$\Delta_{\mathbf{r}}\mathbf{G}^{\mathbf{o}} = \mathbf{A} - \mathbf{B}\mathbf{T}$$

Where A and B are non-zero constants. Which of the following is TRUE about this reaction ?

- (1) Exothermic if B < 0 (2) Exothermic if A > 0 and B < 0
- (3) Endothermic if A < 0 and B > 0 (4) Endothermic if A > 0
- **11.** Adsorption of a gas follows Freundlich adsorption isotherm. In the given plot, x is the mass of

the gas adsorbed on mass m of the adsorbent at pressure p.  $\frac{x}{m}$  is proportional to



12. An organic compound is estimated through Dumas method and was found to evolve 6 moles of CO<sub>2</sub>. 4 moles of H<sub>2</sub>O and 1 mole of nitrogen gas. The formula of the compound is (1)  $C_{12}H_8N$  (2)  $C_{12}H_8N_2$  (3)  $C_6H_8N$  (4)  $C_6H_8N_2$ 

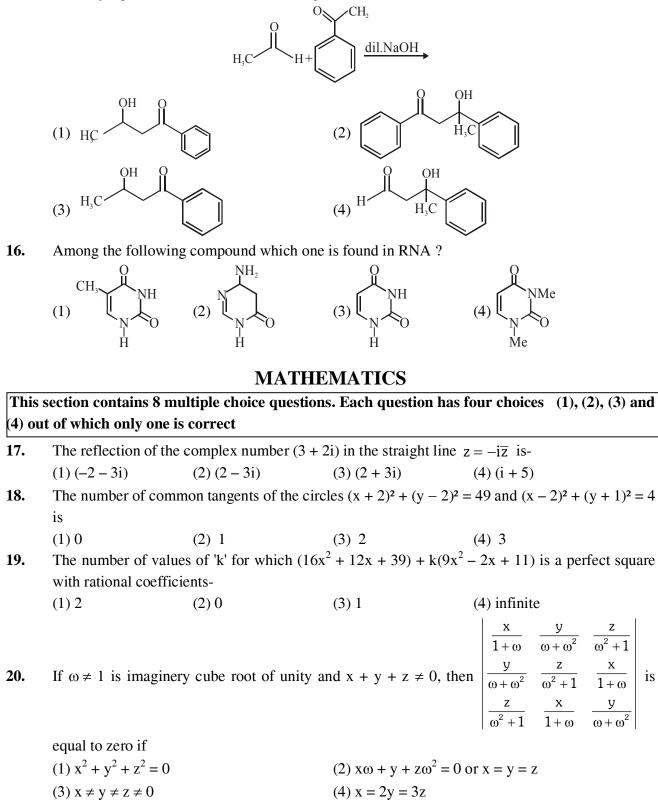
**13.** The coordination number of Th in K<sub>4</sub> [Th(C<sub>2</sub>O<sub>4</sub>]<sub>4</sub> (OH<sub>2</sub>)<sub>2</sub>] is :- $(C_2O_4^{2-} = Oxalato)$ 

14. The relative stability of +1 oxidation state of group 13 elements follows the order :-

- (1) Al < Ga < Tl < In (2) Tl < In < Ga < Al
- (3) Al < Ga < In < Tl (4) Ga < Al < In < Tl

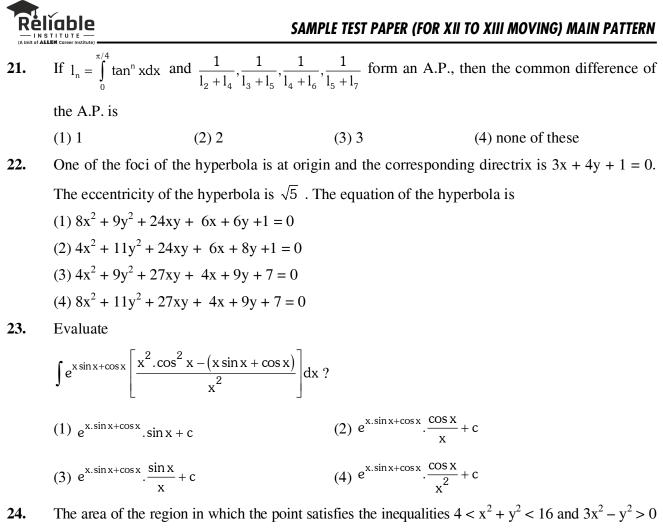


**15.** The major product formed in the following reaction is:



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is:

(1)  $\pi$  (2)  $2\pi$  (3)  $4\pi$  (4)  $8\pi$ 



Space for rough work



## **R-NET (SAMPLE PAPER) (XII to XIII moving students)**

# **ANSWER KEY**

Que.	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20
Ans.	1	3	3	1	3	3	3	3	1	4	4	4	2	3	1	3	1	2	2	2
Que.	21	22	23	24																
Ans.	1	2	2	4																

**R-NET (2020)**